

## The Interaction of 1-Fluoro-2,4-Dinitrobenzene with Amino-Phospholipids in Membranes of Intact Erythrocytes, Modified Erythrocytes, and Erythrocytes Ghosts

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*Summary.* 1-Fluoro-2,4-dinitrobenzene (FDNB) has been used to study the availability of amino-containing phospholipids in erythrocyte membranes and ghosts in an aqueous, isotonic medium. It was found that the addition of bovine serum albumin (BSA) to the medium protects the cells from cation leak and protects some of the amino-phospholipids from reacting with the probe. In isotonic medium without BSA, 46% of the phosphatidylethanolamine and 12% of the phosphatidylserine of erythrocytes and 73% and 21% of these respective lipids of ghosts react with the probe. In the presence of 70  $\mu$ M BSA, 31% of phosphatidylethanolamine and 6.5% of phosphatidylserine of erythrocytes and 59% and 16% of these respective lipids of ghosts react with the probe. The labeling of these lipids does not change under conditions of varying tonicity, or after treatment of erythrocytes with pronase or lysolecithin. The data suggest that 46% of phosphatidylethanolamine and 12% of phosphatidylserine of the erythrocyte membrane are free in a lipid bilayer; 27% and 9% of these respective lipids are loosely bound to proteins which are lost during the preparation of ghosts and 27% of the phosphatidylethanolamine and 79% of the phosphatidylserine are tightly bound to core proteins which remain part of the erythrocyte membrane even after hemolysis.

During the past few years a considerable amount of research has been directed toward determining the structure of cell membranes. Some proteins of the erythrocyte membrane have been isolated and characterized (Maddy, 1970; Marchesi, Steers, Marchesi & Tillack, 1970; Fairbanks, Steck & Wallach, 1971) and attempts have been made to determine their arrangement in this membrane (Rosenberg & Guidotti, 1969; Bender, Garan & Berg, 1971; Bretscher, 1971; Carraway, Kobyłka & Triplett, 1971; Phillips & Morrison, 1971*a, b*). The lipid composition of this membrane has also been extensively studied (Sweely & Dawson, 1969; Cooper, 1970). Although

the gross lipid composition is well established, the arrangement of the lipids in the membrane is almost unknown. Phospholipases A and C have been used to determine the accessibility of the phospholipid molecules in the membrane. The results have shown that the phospholipids of the intact erythrocyte are refractory to these enzymes but that a major portion of the phospholipids in erythrocyte ghosts are hydrolyzed (Roelofsen, Zwaal, Comfurius, Woodward & VanDeenen, 1971; Laster, Sabban & Loyter, 1972).

Nonspecific chemical probes, such as 1-anilino-8-naphthalenesulfonate (ANS), whose fluorescence is altered under conditions of varying hydrophobicity, have been used to determine changes in the conformation of lipids in erythrocyte ghosts perturbed with enzymes (Weidekam, Wallach & Fischer, 1971) or organic solvents (Fortes & Hoffman, 1971; Metcalfe, Metcalfe & Engelman, 1971).

Physical methods have also been used to elucidate the structure of the erythrocyte membrane and it is generally agreed that the phospholipids exist mainly in a bilayer (Chapman, 1972).

Recently, Carraway, Kobyłka, Summers and Carraway (1972) using radioactive acetic anhydride, and Gordesky, Marinetti and Segel (1972) using FDNB have begun to probe certain phospholipids of intact erythrocyte ghosts. These chemical probes react with the amino groups of proteins and the amino-phospholipids such as phosphatidylethanolamine (PE) and phosphatidylserine (PS). Both laboratories have concluded that the structure of the ghost membrane is different from the membrane of the intact erythrocyte.

The present report deals with the reactivity of PE and PS of the erythrocyte membrane with FDNB, under conditions in which the cell is osmotically swollen or crenated, treated with pronase, or altered by incubation with lysolecithin. Evidence that BSA protects erythrocytes as proposed by Furchgott and Ponder (1940) is also presented.

## Materials and Methods

### *Reaction of Cells or Ghosts with FDNB*

The reaction of FDNB with fresh human erythrocytes (0.4 ml packed cells) and erythrocyte ghosts in an isotonic medium, and the isolation and identification of the lipids were carried out as previously described by Gordesky *et al.* (1972).

The blood, obtained by venipuncture and collected in heparinized vacutainers, was centrifuged at 2,000 rpm for 10 min. The plasma and buffy coat were removed and the cells were washed twice with isotonic saline. A sample of 0.4 ml of packed cells, or ghosts made by a modification of the procedure of Dodge, Mitchell and Hanahan (1963) from 0.4 ml of packed cells, were added to 29.6 ml of a reaction medium containing

120 mM  $\text{NaHCO}_3$ , 40 mM  $\text{NaCl}$ , 4.4 mM glucose, 1.5 mM FDNB, and BSA (Sigma-Fraction V) of varying concentrations. The pH of the suspension remained at 8.5 to 8.6 throughout the incubation. The cells (or ghosts) were incubated for 2 hr at 23 °C and then centrifuged. The supernates were discarded and the pellets washed twice with the isotonic bicarbonate buffer.

Ghosts were made from the FDNB-treated cells and the lipids extracted with 20 volumes of  $\text{CHCl}_3/\text{MeOH}$  1:1 (v/v). The lipid extracts were centrifuged and the supernates concentrated under nitrogen for spotting on thin-layer chromatography (TLC) plates. The plates, SG 5763, Merck & Co., Darmstadt, were developed in  $\text{CHCl}_3/\text{MeOH}/\text{H}_2\text{O}$  130:50:7 (v/v). The reacted lipids were identified by their yellow color and by comparison to standards. Unreacted PE and PS were located by staining with ninhydrin. The areas of silica gel containing the specific lipids were scraped into tubes and extracted three times with 2 ml of methanol. The combined extracts were concentrated to 5 ml and the absorbance at 345 nm was determined. Total lipid P was determined by the method of Harris and Papat (1954).<sup>1</sup>

To perform the experiments in a hypotonic medium, 6 ml of a stock FDNB solution (7.5 mM FDNB in 5%  $\text{NaHCO}_3$ ) was added to 39 ml of a solution containing glucose and BSA so that the final concentrations were:  $\text{NaHCO}_3$  80 mM, glucose 3.7 mM, FDNB 1.0 mM, and BSA 46  $\mu\text{M}$ . For the isotonic and hypertonic controls in this experiment, the concentration of  $\text{Na}^+$  was adjusted to 150 mM and 230 mM, respectively. The final osmolarities of the different media were as follows: hypotonic 145 mosm, isotonic 290 mosm, and hypertonic 445 mosm, as determined by a Precision Systems Osmometer Model 2007. The pH of these solutions remained at 8.5 to 8.6 during the incubation with cells.

### *Lysolecithin Treatment*

Chromatographically pure egg lysolecithin (Koch-Light Laboratories, Ltd.) was added to the FDNB stock solution, or to the  $\text{NaHCO}_3$  solution without FDNB, so that the final concentration of lysolecithin in the incubation medium containing the erythrocytes was 40 or 80  $\mu\text{M}$ . The reactions were performed in a medium containing 120 mM  $\text{NaHCO}_3$ , 40 mM  $\text{NaCl}$ , 4.4 mM glucose, 1.5 mM FDNB, and 70  $\mu\text{M}$  BSA.

### *Pronase Digestion*

Washed, packed erythrocytes (0.5 ml) were suspended in 106 mM phosphate buffer, pH 7.4, at a hematocrit of 11% and then reacted with 25 and 100  $\mu\text{g}$  of pronase (Calbiochem, B grade) for varying periods of time as described by Phillips and Morrison (1971b). The cells were centrifuged, washed with buffer and reacted in a medium containing 120 mM  $\text{NaHCO}_3$ , 40 mM  $\text{NaCl}$ , 4.4 mM glucose, 1.5 mM FDNB and 70  $\mu\text{M}$  BSA.

### *Measurement of $\text{Na}^+$ and $\text{K}^+$ Concentration of Erythrocytes*

Erythrocytes were labeled with 1.5 mM FDNB as described above except that BSA was added in varying amounts and the sodium and potassium concentrations of the medium were equimolar to the erythrocyte intracellular electrolytes (sodium = 8 mEq/liter, potassium = 90 mEq/liter). The cells were washed free of FDNB and BSA with

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<sup>1</sup> PS was not always cleanly separated from other phospholipids as was PE. Since the PE/PS molar ratio in normal erythrocytes is reasonably constant (Redd, Swisher, Marinetti & Eden, 1960; Harris & Kellermeyer, 1970), the calculation of total PS in these experiments was based on this ratio and the measured total PE value.

the same buffer free of BSA and FDNB, and then suspended in a Krebs-Henseleit bicarbonate buffer (sodium = 140 mEq/liter, potassium = 5 mEq/liter) for incubation as previously described by Feig, Segel, Shohet and Nathan (1972). Samples were taken at various times for determination of intracellular sodium and potassium by flame photometry (Feig *et al.*, 1972).

## Results

In this study considerable attention will be devoted to the influence of BSA on the labeling of the erythrocyte membrane aminophospholipids by FDNB. Albumin is the major protein in plasma. Consequently, the erythrocytes are constantly surrounded by and interact with albumin. Moreover, albumin is generally known to protect erythrocytes from shape changes and lysis produced by surface forces on glass, and by lytic detergents such as lysolecithin. FDNB was selected as a probe because it is a neutral penetrating probe which rapidly equilibrates between the inside and outside of the cell (Krupka, 1972). FDNB also produces easily measurable yellow derivatives with amino-containing compounds.

The rates of reaction of FDNB with PE and PS, in cells and ghosts, in a medium containing 70  $\mu$ M BSA, are shown in Fig. 1. All reactions

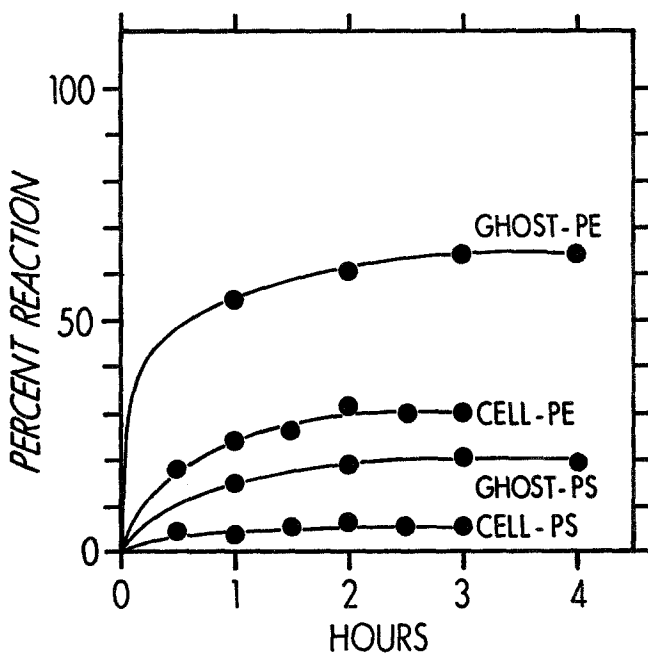


Fig. 1. Rate of reaction of PE and PS of erythrocytes and ghosts with FDNB. The reaction system of erythrocytes and ghosts with FDNB and the analysis of DNB-PE and DNB-PS are explained in the text. The system contained 70  $\mu$ M BSA. The ordinate expresses the percent of the total PE and PS which reacted with FDNB

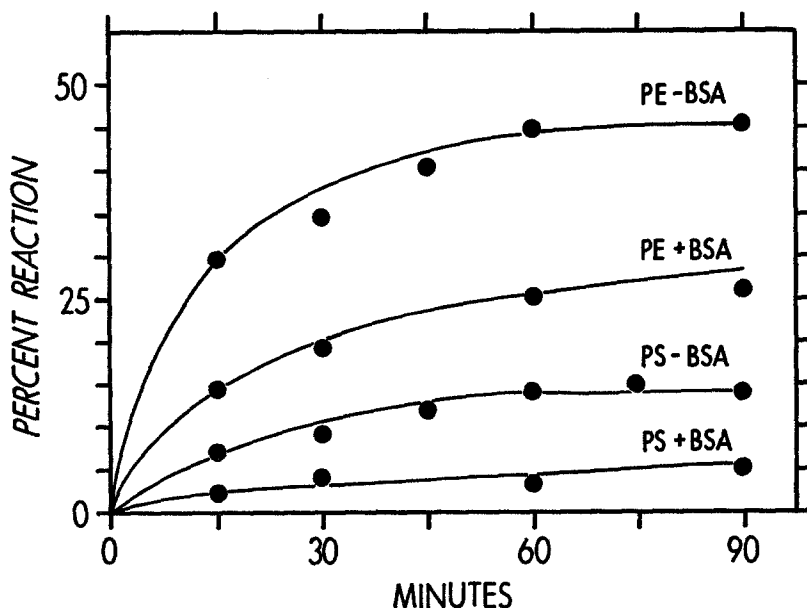


Fig. 2. The effect of BSA on the rate of reaction of PE and PS of erythrocytes with FDNB. The reaction system is explained in the text. When present, the BSA was  $70 \mu\text{M}$ . The ordinate expresses the per cent of total PE and PS which reacted with FDNB

reach a plateau by 2 hr. PE and PS in ghosts react at a faster rate and to a greater extent than do PE and PS in cells. The effect of BSA on the reaction of the amino-phospholipids in cells is shown in Fig. 2. BSA allows less PE and PS to be available for reaction with FDNB. Increasing the concentration of BSA above  $42 \mu\text{M}$  inhibits FDNB labeling of PE and PS in cells and ghosts as seen in Figs. 3 and 4, respectively. About 15% of the PE and 5% of the PS is protected by  $70 \mu\text{M}$  BSA. Although the FDNB labeling of PE and PS in erythrocyte ghosts is greater than that seen with intact cells,  $70 \mu\text{M}$  BSA causes a quantitatively similar decrease in relative FDNB labeling of each amino-phospholipid.

To be certain that the differential reactivity of PE and PS with FDNB in cells or ghosts was not due to a different chemical reactivity of these amino-phospholipids with the reagent, a sonicated dispersion of extracted red cell lipids (free from protein) was reacted with FDNB under the same conditions used for cells. It was found that PE and PS in these sonicated lipid dispersions react nearly to completion with FDNB within 60 min.

BSA decreases the cation permeability induced by FDNB. Increasing concentrations of BSA provide greater protection (Fig. 5). The cells initially contained 8 mEq/liter sodium and 90 mEq/liter potassium. Time zero

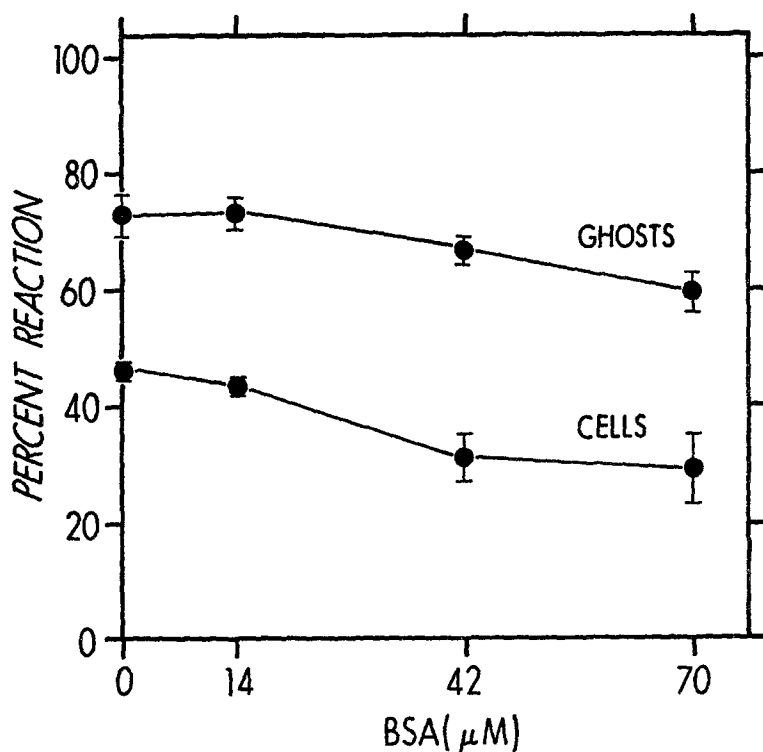


Fig. 3. The effect of BSA on the extent of reaction of PE of erythrocytes and ghosts with FDNB. The reaction of erythrocytes and ghosts with FDNB was carried out as explained in the text using different concentrations of BSA. The ordinate expresses the per cent of total PE which reacted with FDNB. The number of experiments done in duplicate at concentrations of BSA of 0, 14, 42 and 70  $\mu\text{M}$  are, respectively, 4, 2, 2, 6 with ghosts and 2, 3, 7, 7 with cells. Values are mean  $\pm$  SD

values, although taken immediately upon resuspension of treated cells, already show changes in internal sodium in all FDNB-treated cells, and in potassium when BSA was less than 70  $\mu\text{M}$  during the prelabeling period.

In contrast to the protective effect of BSA, the presence of hemoglobin and intracellular proteins from lysed cells has no effect on the FDNB labeling of the membrane amino-phospholipids (Table 1). The relative per cent reaction of PE and PS of ghosts in the presence of their lysates was the same as that of ghosts washed free from these lysate proteins.

The erythrocyte membrane can be modified by incubating cells in hypotonic or hypertonic media so that the cells are swollen or crenated and the membrane is stretched or condensed. Neither of these treatments alters the amount of PE or PS which is available to react with FDNB (Table 2).

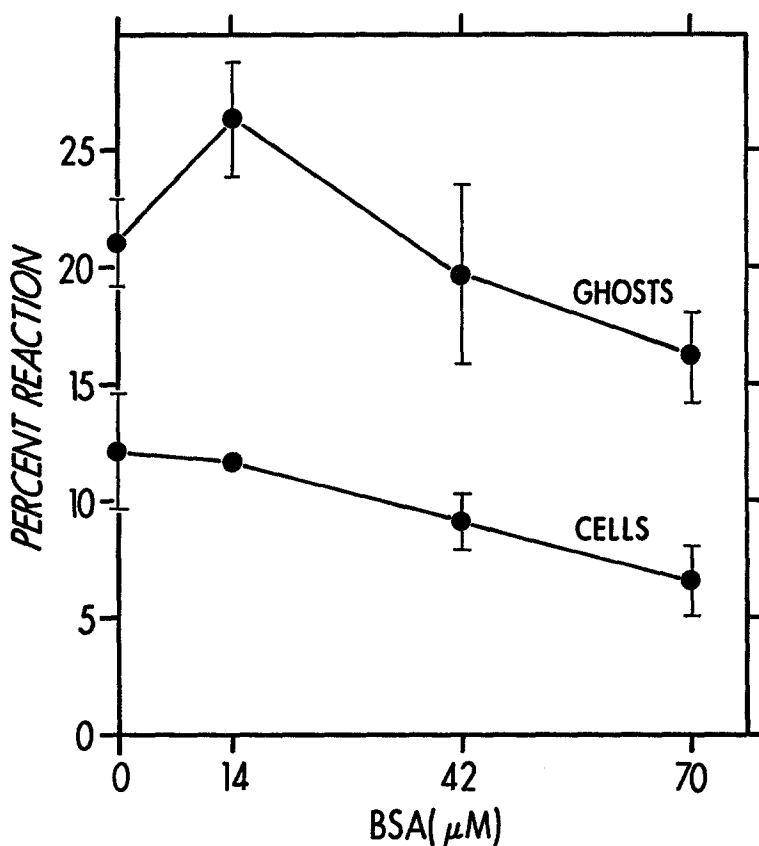


Fig. 4. The effect of BSA on the extent of reaction of PS of erythrocytes and ghosts with FDNB. The reaction of erythrocytes and ghosts with FDNB was carried out as explained in the text using different concentrations of BSA. The ordinate expresses the per cent of total PS which reacted with FDNB. The number of experiments done in duplicate at concentrations of BSA of 0, 14, 42 and 70  $\mu\text{M}$  are, respectively, 2, 2, 2, 7 with ghosts and 2, 1, 5, 7 with cells. Values are mean  $\pm$  SD

Pronase, a nonspecific proteolytic enzyme, was used to digest a portion of the external proteins of the erythrocyte membrane (Phillips & Morrison, 1971 *b*). At enzyme concentrations of 25 and 100  $\mu\text{g}/\text{ml}$  there was no change in the amount of PE which reacts with FDNB (Table 3).

Erythrocyte membranes were also modified with lysolecithin, a surface active agent capable of lysing erythrocytes (Wilbur & Collier, 1943). As seen in Table 4, no appreciable difference in FDNB labeling of PE or PS was found between the control cells and cells reacted in the presence of 40 and 80  $\mu\text{M}$  lysolecithin under conditions where the cells changed shape but no hemolysis occurred. The cells exposed to the lower concentration of

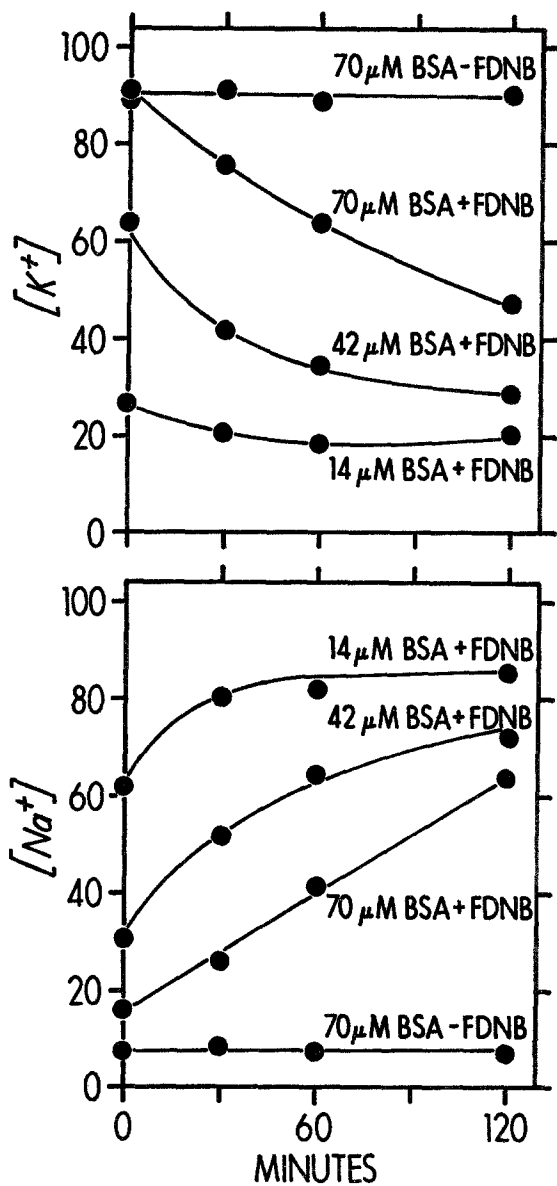


Fig. 5. The effect of BSA on  $Na^+$  and  $K^+$  leak in erythrocytes treated with FDNB. The  $Na^+$  and  $K^+$  leak measurement and the FDNB treatment were done as explained in the text. The values are mEq/liter cells

lysolecithin became crenated (echinocytes) whereas cells exposed to the higher concentration became spherocytes.

Erythrocyte hemolysis serves as an indicator of the membrane permeability to large molecules. Less than 2% hemolysis was observed in the hyper-



Table 1. The effect of the cell lysate on the per cent of total PE and PS of erythrocytes which reacts with FDNB<sup>a</sup>

System	PE	PS
I Intact cells	30.5 $\pm$ 4.1	8.4 $\pm$ 2.2
II Lysed cells with Hb present	52.8 $\pm$ 4.8	14.0 $\pm$ 2.0
III Washed ghosts	52.8 $\pm$ 4.5	12.2 $\pm$ 3.2

<sup>a</sup> The reactions were carried out in an isotonic medium as given in the text. The FDNB concentration was 1.5 mM. The BSA concentration was 70  $\mu$ M. The values represent the mean  $\pm$  SD of duplicate analyses of four experiments. System I represents the normal reaction. In system II the cells were lysed and the reaction carried out with the ghosts and the lysate both present. In system III the cells were lysed and the ghosts washed once with the lysing medium (5 mM Tris, 1 mM EDTA, pH 7.4).

Table 2. Per cent of total PE and PS of intact erythrocytes which reacts with FDNB in media of different tonicity<sup>a</sup>

Medium	PE	PS
Hypotonic medium	25.1 $\pm$ 2.7	5.4 $\pm$ 1.2
Isotonic medium	27.5 $\pm$ 1.6	8.0 $\pm$ 1.6
Hypertonic medium	30.7 $\pm$ 5.9	6.2 $\pm$ 1.3

<sup>a</sup> The systems contained 0.4 ml packed erythrocytes, 1.0 mM FDNB, 46  $\mu$ M BSA, 80 mM NaHCO<sub>3</sub>, 3.7 mM glucose and varying amounts of NaCl as follows: hypotonic medium — none, isotonic medium — 70 mM NaCl, hypertonic medium — 150 mM NaCl. The values are the mean  $\pm$  SD of duplicate analyses of two experiments.

Table 3. Effect of pronase on the per cent of total PE of erythrocytes which reacts with FDNB<sup>a</sup>

Time of Reaction (min)	Per cent reaction
0	29.9
15	28.2
30	28.9
60	28.5

<sup>a</sup> Erythrocytes were treated with pronase, washed and then reacted with FDNB as described in the text. Values are the average of two numbers which varied no more than 1.5% from the mean.

tonic, isotonic, or hypotonic media during the reaction of the erythrocytes with FDNB in the presence of 70  $\mu$ M BSA for 4 hr at room temperature. After resuspension of the FDNB-treated cells for 6 more hours in hypertonic, isotonic or hypotonic media containing 3.7 mM glucose, there is less than

Table 4. Effect of lysolecithin on the per cent of total PE and PS of erythrocytes which reacts with FDNB<sup>a</sup>

Lysolecithin conc. ( $\mu$ M)	PE	PS
0	$29.4 \pm 3.4$	$8.0 \pm 2.1$
40	$27.0 \pm 0.4$	$7.8 \pm 1.8$
80	$27.1 \pm 3.4$	$7.2 \pm 2.8$

<sup>a</sup> Erythrocytes were reacted with FDNB in the presence of lysolecithin and BSA as described in the text. Values are the mean  $\pm$ SD of four experiments.

2% hemolysis. Only after 24 hr does extensive (25 to 35%) hemolysis occur. Therefore, in these studies which utilize a completely aqueous medium, the erythrocytes are stable during the important 2-hr period of exposure to FDNB, so that this probe is interacting with an intact membrane.

### Discussion

FDNB freely permeates the erythrocyte membrane and dinitrophenylates protein and lipid amino groups (Krupka, 1972). Krupka provided evidence that FDNB rapidly equilibrates between the inside and outside of the erythrocyte. In these studies we examined the reaction of amino-phospholipids of the erythrocyte membrane with FDNB. The reaction of FDNB with membrane PE and PS of intact cells was maximal in 2 hr (Gordesky *et al.*, 1972) and caused less than 2% hemolysis during this time. Using very different conditions of pH and hematocrit, Poensgen and Passow (1971) found, in three different experiments, 10 to 47% of PE and no PS of erythrocytes reacts with FDNB.

Erythrocyte membrane proteins, intracellular proteins and added BSA also react with FDNB. To insure that the plateau observed in Fig. 1 was not due to the exhaustion of FDNB, the yellow supernatant from an initial labeling experiment was reacted with fresh erythrocytes and the lipids were extracted in the usual way. Additional DNB-PE was demonstrated by TLC, thus showing that the plateau is not due to exhaustion of FDNB but rather is due to saturation of available PE and PS in the membrane. Intracellular proteins, primarily Hb, did not alter the FDNB labeling of PE and PS of ghosts (Table 1). Hemoglobin, therefore, does not act as a sink for FDNB in this system and FDNB remains in excess. Moreover, unlike albumin, the intracellular proteins do not protect PE and PS from reaction with FDNB.

A calculation can be made of total amino groups in the incubation mixture with intact erythrocytes to show that FDNB labels but a small number of these groups. The amount of protein and lipid in 0.4 ml of packed erythrocytes can be estimated from the values of Harris and Kellermeyer (1970) as follows: PE + PS—0.5  $\mu$ moles, membrane protein<sup>2</sup>—0.002  $\mu$ moles, hemoglobin—2.2  $\mu$ moles. The maximum amount of BSA added was 2.1  $\mu$ moles. Therefore, a total of 4.3  $\mu$ moles of protein are available to react with FDNB. Each hemoglobin molecule has 80 free amino groups. The total hemoglobin in this reaction contains 175  $\mu$ moles of free amino groups. The 2.1  $\mu$ moles of BSA contribute approximately 103  $\mu$ moles of free amino groups. Membrane protein and nonhemoglobin intracellular protein<sup>3</sup> contain approximately 0.15  $\mu$ moles of free amino groups and PE and PS contribute 0.5  $\mu$ moles of free amino groups. Thus, a total of approximately 279  $\mu$ moles of free amino groups exist in the reaction medium. Since 45  $\mu$ moles of FDNB are added and excess FDNB is present when the reaction has reached a plateau, these results demonstrate that most of the amino groups (as well as SH, OH, tyrosyl, guanidinium and histidyl groups) which could potentially react with FDNB are masked and not available to FDNB.

When FDNB is used as a probe for the amino groups of PE and PS in the intact erythrocyte membrane, 46% of the PE and 12% of the PS amino groups are labeled in contrast to 73% and 21%, respectively in ghosts<sup>4</sup>. It is clear that in the preparation of ghosts an alteration has occurred in the membrane which leads to an increase in the availability of PE and PS for FDNB.

Evidence is available that when ghosts are prepared from intact red cells an alteration of the membrane occurs. This is supported by the observations of Roelofson *et al.* (1971) and Laster *et al.* (1972), that phospholipids in the membrane of intact erythrocytes are not attacked by phospholipase A or phospholipase C but in ghosts these phospholipids are readily attacked.

Moreover, proteins of ghosts are extensively hydrolyzed by trypsin whereas proteins of the intact erythrocyte are not extensively hydrolyzed (Triplett & Carraway, 1972).

In the presence of 70  $\mu$ M BSA, intact erythrocytes and ghosts exhibit a similar relative decrease in PE and PS labeling. The 46% of reactive PE in intact cells is reduced to 31% and the 73% of reactive PE in ghosts is

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2 This calculation assumes an average molecular weight of 100,000.

3 This calculation assumes 75 amino groups per 100,000 mol wt.

4 Values previously published in an abstract (Gordesky & Marinetti, 1972) were high due to an error in the slope of the standard curve for DNB-PE.

reduced to 59% by BSA. The reactive PS in erythrocytes is reduced from 12% to 6.5% and in ghosts it is reduced from 21% to 16%. Therefore, the decrease in reactivity of PE in both intact cells and ghosts caused by BSA is 14 to 15%. Furthermore, the decrease in reactivity of PS in both intact cells and ghosts caused by BSA is 5%. These data strongly suggest not only that BSA has a protective effect on PE and PS but that BSA protects the PE and PS at specific membrane sites present in both the intact cells and ghosts.

The protective action of BSA might be due to its binding to the membrane in such a way that some of the polar groups of PE and PS are masked. It is noteworthy that Ponder (1948) has demonstrated the binding of albumin to the erythrocyte. He postulated that albumin became embedded in the membrane and formed a specific part of the membrane structure.

The FDNB-induced change in cation permeability of erythrocytes has been described by Poensgen and Passow (1971). We have confirmed and extended their finding that cation leak is increased by FDNB. Added BSA protected the cells from the FDNB-induced cation leak, possibly due to the preservation of positively charged amino groups of PE and PS in the membrane.

When the cell membrane of intact cells was stretched in a hypotonic medium or condensed in a hypertonic medium, no significant change in the reactivity of PE and PS with FDNB was seen. When the membrane was altered by lysolecithin to cause sphering and crenation (Lichtman & Marinetti, 1972), or by pronase, which partially degrades but does not release proteins from the erythrocyte surface (Phillips & Morrison, 1971*b*), there was no effect on the FDNB labeling of PE. Albumin protected the cells from lysis by lysolecithin. This effect of albumin is twofold. Albumin binds lysolecithin thus making less lysolecithin free to interact with the red cell membrane. Albumin also binds to the cell membrane giving it partial protection from lysis by lysolecithin and protecting some amino-phospholipids from reacting with FDNB.

We suggest since 46% of the PE molecules and 12% of the PS molecules of the intact erythrocyte react freely with FDNB, they are not associated with protein and are in a lipid bilayer in the membrane. One-third of these PE molecules and one-half of these PS molecules may be protected by BSA. Moreover, in ghosts an additional 27% of the PE molecules and an additional 9% of the PS molecules are made available to FDNB. We postulate that these newly exposed amino-phospholipids are associated loosely with proteins which are lost during the preparation of these ghosts (Dodge *et al.*, 1963). The remaining 27% of the PE and 79% of the PS molecules which

are refractory to FDNB in ghosts are considered to be strongly bound to core proteins which remain part of the membrane structure even after hemolysis. These results and postulates are consistent with the Singer and Nicolson (1972) model of membrane structure.

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